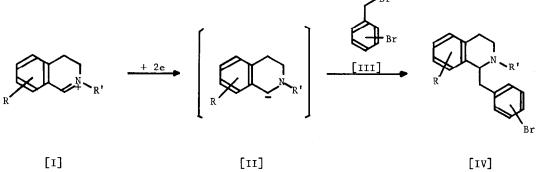
ELECTROREDUCTIVE SYNTHESIS OF 1-(BROMOBENZYL)-ISOQUINOLINE DERIVATIVES AND ITS APPLICATION TO CULARINE SYNTHESIS¹

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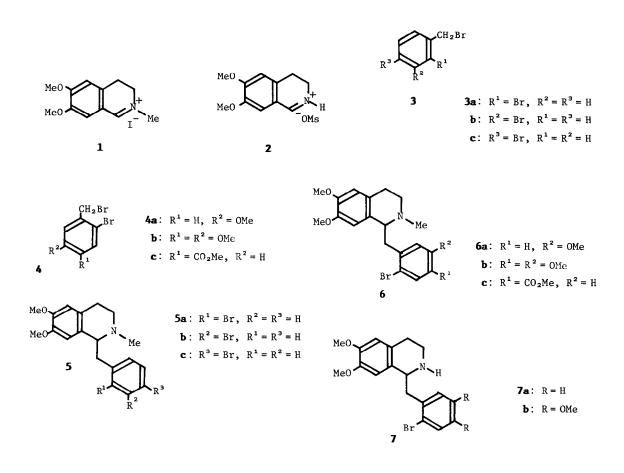
Electrochemical reduction of immonium salts in the presence of bromobenzylbromide derivatives gave l-(bromobenzyl)-isoquinoline derivatives in moderate yields. This reaction is useful in the synthesis of several natural alkaloids as exemplified in the synthesis of Cularine.

Although 1-(bromobenzyl)-isoquinoline derivatives are useful intermediates in the synthesis of Cularine,² Aporphine,³ Dibenzopyrocoline,⁴ Berbine,⁵ and Bisbenzylisoquinoline,⁶ their synthesis has not necessarily been established. The synthetic methods hitherto known are mainly the bromination of the benzyl moiety of 1-benzyl-isoquinoline derivatives, and hence the reaction is rather lack of selectivity and generality.

In the present study, a new and simple method of 1-(bromobenzyl)-isoquinoline synthesis through electroreduction⁷ of immonium salts $[I]^8$ and bromobenzylbromides $[III]^9$ at - 1.8 V vs. SCE is described.



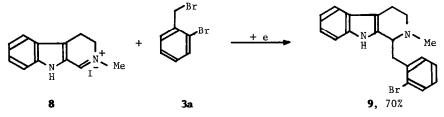
Our new method is highly versatile as summarized in Table I.



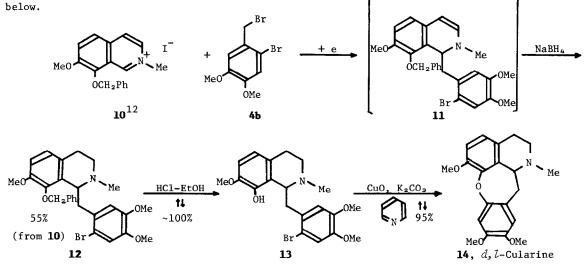
Immonium Salts	Bromobenzylbromides	Products	Yields (%)
1	3a	5a	65
1	3ь	5ь	47
1	3c	5e	48
1	4a	ба	61
1	4Ъ	6Ъ	51
1	4c	бс	60
2	3a	7a	50
2	4b	7ъ	53

It is noteworthy that the anions [II] generated from immonium salts [I] react selectively with the bromomethyl moiety of the bromobenzylbromide [III].

Furthermore, our method is applicable to indole derivatives as well as isoquinoline derivatives.



A new synthesis of Cularine could be performed by using this simple method as depicted



The electroreduction of an immonium salt $(10)^{7a}$ at - 2.0 V vs. SCE in the presence of a bromobenzylbromide (4b) followed by NaBH, reduction of the resulting enamine (11) afforded a product 12. Debenzylation of 12 to a phenolic compound 13 in ethanolic hydrochloric acid solution¹³ and subsequent Ullmann reaction¹⁴ by CuO in pyridine gave *d*, *l*-Cularine (14), ¹⁴ the total yield from 10 being 52%.

Although yields of our new benzylation reaction are not always satisfactory, the simplicity and versatility of the electrochemical reaction make this new synthesis promising.

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References and Notes

1) Electroorganic Chemistry. 51.

- 2) a) T. Kametani, K. Fukumoto, and M. Fujihara, *Chem. Pharm. Bull.*, 20, 1800 (1972).
 b) H. Iida, H. C. Hsu, and T. Kikuchi, *ibid.*, 21, 1001 (1973).
- 3) a) A. H. Jackson and J. A. Martin, J. Chem. Soc. (C), 1966, 2061.
 - b) T. Kametani, S. Shibuya, K. Kigasawa, M. Hiiragi, and O. Kusama, *ibid.*, 1971, 2712.
- 4) a) T. Kametani and K. Ogasawara, J. Chem. Soc. (C), 1967, 2208.
 - b) F. Benington and R. D. Morin, J. Org. Chem., 32, 1050 (1967).
- 5) a) T. Kametani and M. Ihara, J. Chem. Soc. (C), 1967, 530.
 - b) C. Tani, S. Takao, H. Endo, and E. Oda, J. Pharm. Soc. Jpn., 93, 268 (1973).
- 6) a) T. Kametani, H. Iida, and K. Sakurai, J. Chem. Soc. (C), **1971**, 1024.
 - b) R. W. Doskotch, J. D. Phillipson, A. B. Ray, and J. L. Beal, J. Org. Chem., 36, 2409 (1971).
- 7) a) T. Shono, K. Yoshida, K. Ando, Y. Usui, and H. Hamaguchi, Tetrahedron Lett., 1978, 4819.
 - b) T. Shono, Y. Usui, T. Mizutani, and H. Hamaguchi, *ibid.*, 21, 3073 (1980).
- 8) Immonium salts were prepared according to the established method.
- 9) 2-Bromo-5-methoxybenzylbromide (4a) was prepared according to the reported method.¹⁰ Other derivatives of bromobenzylbromides (3a-c, 4b,c) were synthesized by bromination of their bromotoluene derivatives¹¹ by NBS.
- 10) R. Breslow, S. Garratt, L. Kaplan, and D. LaFollette, J. Am. Chem. Soc., 90, 4051 (1968)
- 11) The starting bromotoluene derivatives were prepared according to the reported method.
 4b. T. Heap, T. G. H. Jones, and R. Robinson, J. Chem. Soc., 1927, 2021.
 4c. A. M. Fleifel, J. Org. Chem., 25, 1024 (1960).
- 12) A. H. Jackson, G. W. Stewart, G. A. Charnock, and J. A. Martin, J. Chem. Soc., Perkin Trans 1, 1974, 1911.
- 13) H. Iida, T. Kikuchi, K. Sakurai, and T. Watanabe, J. Pharm. Soc. Jpn., 89, 645 (1969).
- 14) S. Ishiwata, T. Fujii, N. Miyaji, Y. Satoh, and K. Itakura, Chem. Pharm. Bull., 18, 1850 (1970).

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